



# Lattice dynamics and anharmonicity of CaZrF<sub>6</sub> from Raman spectroscopy and *ab initio* calculations



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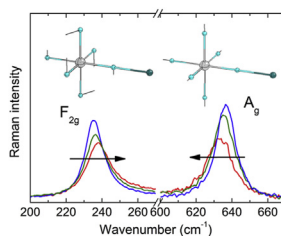
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## HIGHLIGHTS

- A Raman and *ab initio* study of the lattice dynamics of CaZrF<sub>6</sub> was performed.
- All the Raman-active modes expected on the basis of the group theory were identified.
- The temperature-dependence of the CaZrF<sub>6</sub> Raman frequencies follows an unusual trend.
- Explicit anharmonicity dominates for both F<sub>2g</sub> and A<sub>g</sub> Raman modes.
- The NTE of CaZrF<sub>6</sub> cannot be accurately predicted by the quasi-harmonic approximation.

## GRAPHICAL ABSTRACT



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## ABSTRACT

Very recently it has been found that CaZrF<sub>6</sub> exhibits a very large and isotropic negative thermal expansion (NTE), even greater than the current most popular NTE materials. In this work, the vibrational dynamics of CaZrF<sub>6</sub> has been investigated by temperature-dependent Raman spectroscopy combined with *ab initio* calculations. As expected on the basis of the group theory for CaZrF<sub>6</sub>, three Raman-active modes were identified: the F<sub>2g</sub> mode peaked at about 236 cm<sup>-1</sup>, the E<sub>g</sub> mode at around 550–555 cm<sup>-1</sup>, and the A<sub>g</sub> mode peaked at about 637 cm<sup>-1</sup>. The temperature dependence of their frequencies follows an unusual trend: the F<sub>2g</sub> mode, due to bending vibrations of fluorine atoms in the linear Ca-F-Zr chain, is hardened with increasing temperature, while the A<sub>g</sub> mode, corresponding to Ca-F-Zr bond stretching vibrations, is softened. We explain this anomalous behavior by separating implicit and explicit anharmonicity for both F<sub>2g</sub> and A<sub>g</sub> modes. In fact, cubic anharmonicity (three-phonon processes) is observed to dominate the higher-frequency A<sub>g</sub> phonon-mode, quartic anharmonicity (four-phonon processes) is found to dominate the lower-frequency F<sub>2g</sub> phonon-mode. As a result, the large NTE of CaZrF<sub>6</sub> cannot be accurately predicted through the quasi-harmonic approximation.

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## 1. Introduction

Most materials expand upon heating. In the last two decades, after the discovery of materials that display strong negative thermal expansion (NTE) over wide temperature ranges, such as ZrW<sub>2</sub>

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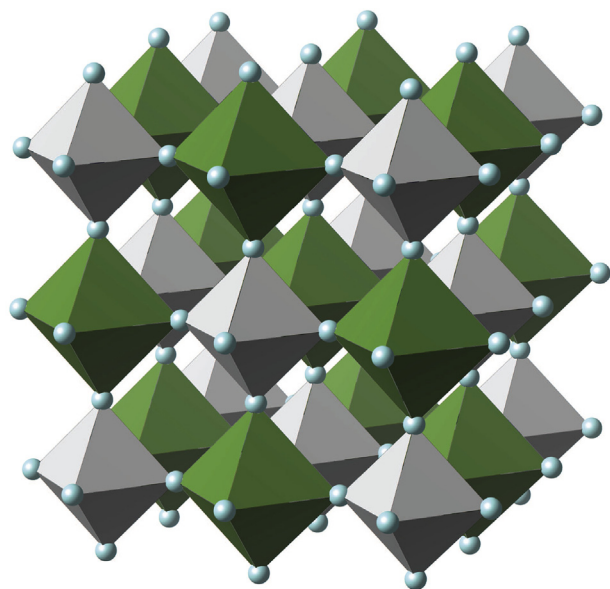
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$O_8$ ,  $Zn(CN)_2$ , metal-organic frameworks and more recently  $ScF_3$  [1–6], the interest in NTE has rapidly increased. NTE phenomenon is known to arise from a range of different physical mechanisms, such as ferroelectricity and magnetic transitions [7–9], valence transitions in the case of intermetallic and fulleride materials [10,11], low-energy vibrational modes for open-framework solids [12–15]. In particular, it is now accepted that the mechanism driving NTE in the framework structures involves transverse vibrational displacement of the central linking atom [13–16], but a general understanding of the physical origin of NTE (and related phenomena) is still lacking.

Very recently it has been discovered that  $CaZrF_6$  displays a very large and isotropic NTE (up to  $-18 \times 10^{-6} K^{-1}$  at 100 K) [18], even greater than the current most popular NTE materials like  $ZrW_2 O_8$  or  $ScF_3$ . In addition, the NTE temperature range of  $CaZrF_6$  ranges from below 10 K–1173 K or even higher, thus making this material until now the one with the largest NTE temperature range. Besides this,  $CaZrF_6$  is transparent over a wide wavelength range, can be fabricated in ceramic form and handled in air, thus being beneficial for the design of controlled thermal expansion materials and for technological applications.

From the structural point of view,  $CaZrF_6$  consists of corner-shared  $CaF_6$  and  $ZrF_6$  octahedra (Fig. 1) and maintains a high symmetry (space group,  $Fm-3m$ ) among the whole temperature range. In sharp contrast with  $ZrW_2 O_8$ , no phase transition takes place in  $CaZrF_6$ . Besides, the metal atoms both locate in the special crystallographic sites (Ca (0,0,0) and Zr (0.5,0.5,0.5)) and all the fluorine atoms share the same chemical environment. The prevailing NTE characteristic and its high cubic symmetry structure enable  $CaZrF_6$  to be an excellent candidate for the study of NTE mechanism, whose knowledge would facilitate the development of specifically tailored materials based on a priori knowledge rather than empirical or casual procedures.

Because very limited information are available concerning the vibrational properties of  $CaZrF_6$  and, even better, about its NTE mechanism, in this work we aim to investigate the lattice dynamics of  $CaZrF_6$  by temperature-dependent Raman scattering combined



**Fig. 1.** Sketch of the crystal structure of  $CaZrF_6$  [17]: gray and green octahedra are  $ZrF_6$  and  $CaF_6$  units, respectively, cyan spheres are fluorine atoms. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

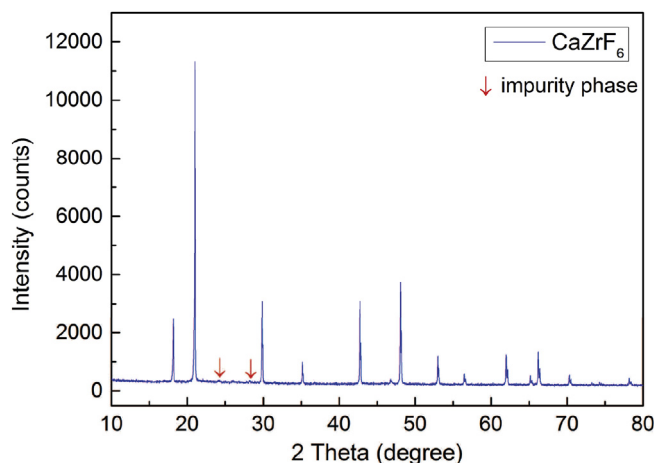
with *ab initio* calculations. Raman spectroscopy only probes zone center modes thus providing a limited view of the lattice dynamics, nevertheless we can get new insights into the vibrational dynamics of  $CaZrF_6$  related to the NTE.

## 2. Experimental and computational details

The compound of  $CaZrF_6$  was prepared by a solid state synthesis method. The precursors of high purity (99.99%)  $CaF_2$  and  $ZrF_4$  powders, were mixed with equal molar ratio. Then the mixture was pressed into a small pellet with diameter of  $\sim 5$  mm and then sandwiched into  $NH_4 F$  powder to be pressed into a bigger one with diameter of  $\sim 10$  mm. The pellet was loaded into Cu tube with two ends sealed and transferred to a muffle furnace immediately, followed by heating them at  $700^\circ C$  for 1 h, before they were slowly decreased to room temperature. Lab X-ray powder diffraction (XRD) data of  $CaZrF_6$  was collected at room temperature, operating at 40 kV and 40 mA with  $Cu K\alpha_{1/2}$  radiation (PANalytical, PW 3040-X'Pert Pro). The XRD data shows that the main phase is  $CaZrF_6$  with an extremely tiny amount of  $ZrO_2$  impurities in monoclinic phase (Fig. 2). Structural refinement gives a lattice parameter of  $8.4630 \text{ \AA}$  with fluorine cell position  $x = 0.2366$ , in agreement with Hancock et al. [18].

Unpolarized micro-Raman spectra were collected in backscattering geometry from  $CaZrF_6$  sample using a triple monochromator (Horiba-Jobin Yvon, model T64000), set in double-subtractive/single configuration, and equipped with holographic gratings having 1800 lines/mm. The Raman spectra were excited by the 514.5 nm laser line focused onto a spot of about  $2 \mu m$  in size through the lens of a  $50 \times$  microscope objective ( $NA = 0.5$ ). The laser power on the sample surface was kept below 10 mW, and the scattered radiation, filtered by the fore double monochromator, was detected at the spectrograph output by a charge-coupled-device detector with  $1024 \times 256$  pixels, cooled by liquid nitrogen. The spectral resolution was better than  $0.6 \text{ cm}^{-1}/\text{pixel}$  while an accurate wave-number calibration of the spectrometer was achieved based on the emission lines of a Ne spectral lamp. The temperature dependence of the Raman spectra of  $CaZrF_6$  was investigated in the temperature range between room temperature and  $200^\circ C$  with the use of an electric heating plate.

*Ab initio* calculations based on density functional theory in the framework of local density approximation have been performed to calculate the Raman active vibrational modes of  $CaZrF_6$ . This



**Fig. 2.** X-ray diffraction pattern of  $CaZrF_6$  with a tiny amount of  $ZrO_2$  impurity marked by red arrows. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

represents an invaluable help for the assignment of the observed Raman modes. The CRYSTAL-14 package was used in the present calculations, a periodic *ab initio* code that uses a Gaussian-type basis set to represent the crystalline orbitals [19]. All-electron basis sets were employed, with 86-511d3G contraction (one s, four sp, and one d shell) for calcium [20], 97-636121 contraction (one s, three sp, one d, one sp, and two d shells) for zirconium [21], and with 7-311G contraction (one s and three sp shells) for fluorine [22]. Pure density functional theory calculations have been carried out using von Barth-Hedin [23] exchange and correlation functionals. The Brillouin zone was sampled by a  $12 \times 12 \times 12$  Monkhorst-Pack k-points mesh [24], while the truncation criteria for bi-electronic integrals (Coulomb and exchange series), controlled by five parameters, was set to default values (7 7 7 7 14). In order to ensure good convergence, the self-consistent-field convergence threshold on total energy was set to  $10^{-8}$  Hartree for geometry optimization, to  $10^{-10}$  Hartree for frequency calculation. The vibrational frequencies at the  $\Gamma$ -point of the Brillouin zone were calculated in the harmonic approximation by diagonalizing the mass-weight Hessian matrix, whose  $(i,j)$  element is defined as  $W_{ij} = H_{ij}/M_i M_j$ , with  $M_i$  and  $M_j$  masses of the  $i$ th and  $j$ th atoms, respectively. A more detailed description of the computational aspects can be found in the refs. [19,25,26].

### 3. Results and discussion

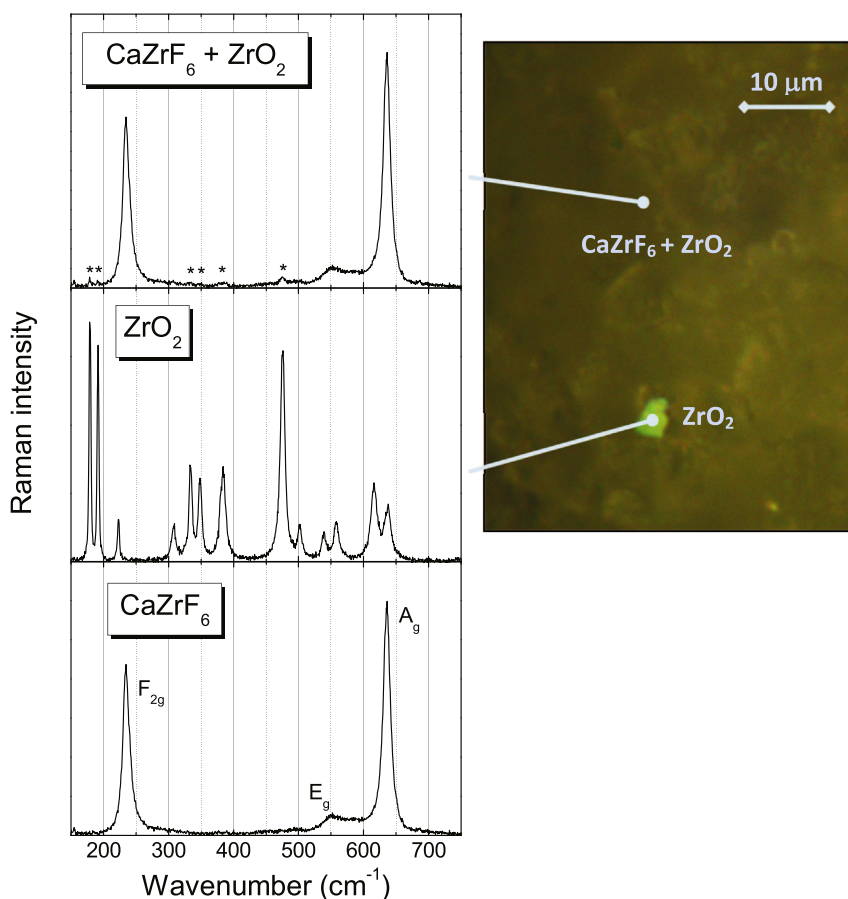
On the basis of the group factor analysis, 21 optical modes of  $\text{CaZrF}_6$  at the  $\Gamma$ -point of the Brillouin zone are expected, with the

following irreducible representation of the normal vibrational modes:

$$3F_{1u} + 1F_{2u} + 1F_{2g} + 1F_{1g} + 1E_g + 1A_g \quad (1)$$

where the  $F_{1u}$ ,  $F_{2u}$ ,  $F_{1g}$  and  $F_{2g}$  modes are triple degenerated, the  $E_g$  mode is double degenerated. The three modes  $F_{2g}$ ,  $E_g$  and  $A_g$  are Raman active, whereas the three  $F_{1u}$  modes are infrared active and the remaining  $F_{1u}$  and  $F_{1g}$  modes are inactive (silent). In order to discriminate the Raman  $F_{2g}$ ,  $E_g$  and  $A_g$  modes, it would be necessary to control the polarization direction of the incident and scattered light with respect to the crystallographic axes. However this is impossible in our powder sample because no crystallographic axis can be identified. Accordingly, for the assignment of the observed Raman modes, we have calculated the vibrational frequency of the Raman-active modes through *ab initio* quantum-mechanical methods (see below).

The unpolarized Raman spectrum collected in our  $\text{CaZrF}_6$  sample at room temperature is shown in the top panel of Fig. 3. Two intense Raman bands are observed at about 236 and 637  $\text{cm}^{-1}$ , as well as a broad band at about 550–555  $\text{cm}^{-1}$ . Moreover, we notice a set of weak Raman bands at ~179, 191, 324, 348, 384, and 475  $\text{cm}^{-1}$ , marked by asterisk in the top panel of Fig. 3, which correspond to the Raman peak positions of monoclinic  $\text{ZrO}_2$  [27,28], tiny impurity in the sample previously detected by x-ray diffraction (Fig. 2). In this regard, under direct optical inspection of a color camera interfaced to the microscope objective used to collect the scattered radiation, we have identified in the sample a grain of  $\text{ZrO}_2$  impurity



**Fig. 3.** Room temperature unpolarized Raman spectrum of: i)  $\text{CaZrF}_6$  with a tiny amount of  $\text{ZrO}_2$  impurity (top panel, the asterisks mark the most intense Raman peaks of  $\text{ZrO}_2$ ) ii) a grain of  $\text{ZrO}_2$  impurity identified in the sample (middle panel and picture on the right side) iii)  $\text{CaZrF}_6$  after partial removal of the  $\text{ZrO}_2$  Raman contribution (bottom panel).

**Table 1**

Observed and calculated vibrational frequencies of CaZrF<sub>6</sub> (in cm<sup>-1</sup>) at the  $\Gamma$ -point of Brillouin zone. The Grüneisen parameters  $\gamma_i = -V/\omega_i(\partial\omega_i/\partial V)$  of each vibrational mode were calculated by pressure-dependent *ab initio* calculations (see text).

Modes	Expt.	Theory	$\gamma_i$
Raman modes			
F <sub>2g</sub>	236	251	-0.36
E <sub>g</sub>	~550	529	1.69
A <sub>g</sub>	637	637	1.26
Infrared modes			
F(1)1u		193	-1.83
F(2)1u		313	2.10
F(3)1u		542	1.68
Silent modes			
F <sub>1g</sub>		41	-34.9
F <sub>2u</sub>		117	-3.41

(picture in Fig. 3) whose Raman spectrum is reported in the middle panel of Fig. 3. Then, after proper normalization, the Raman contribution of ZrO<sub>2</sub> has been subtracted, thus obtaining the Raman spectrum of CaZrF<sub>6</sub> shown in bottom panel of Fig. 3.

Theoretical calculations help to assign the observed Raman peaks of CaZrF<sub>6</sub> and to gain further insight into its vibrational properties. First of all, geometry optimization was performed leading to the cell parameter  $a = 8.4260$  Å and to the fluorine cell position  $x = 0.2389$ . These structural parameters, which differ from the experimental values less than 1.0%, have been employed in the subsequent frequency calculation. Table 1 shows the predicted vibrational frequencies of CaZrF<sub>6</sub> at the  $\Gamma$ -point of the Brillouin zone. To the best of our knowledge, no experimental data are available for both Raman and infrared vibrational modes of CaZrF<sub>6</sub>, with the exception of the present Raman study. Thanks to the direct comparison between calculated and experimental frequencies (see Table 1), we can identify the Raman bands of CaZrF<sub>6</sub> shown in Fig. 3. Specifically, the Raman band observed at ~236 cm<sup>-1</sup> is assigned to the F<sub>2g</sub> symmetry mode, the band at ~550 cm<sup>-1</sup> corresponds to the E<sub>g</sub> symmetry mode, the band at ~637 cm<sup>-1</sup> is the A<sub>g</sub> symmetry mode. From the eigenvectors analysis (Fig. 4), the F<sub>2g</sub> vibrational mode corresponds to bending vibrations of fluorine atoms perpendicular to the linear Zr-F-Ca bonds, leading to twisting of the ZrF<sub>6</sub> and CaF<sub>6</sub> octahedra. Instead the E<sub>g</sub> and A<sub>g</sub> vibrational modes correspond, respectively, to the asymmetric and symmetric stretching of the Zr-F and Ca-F bonds in the ZrF<sub>6</sub> and CaF<sub>6</sub>

octahedra. In particular, the A<sub>g</sub> mode is a breathing mode, as depicted in Fig. 4.

Temperature dependence of vibrational modes plays a key role for the negative thermal expansion of CaZrF<sub>6</sub>. For that reason, we have investigated the temperature dependence of the frequency of the F<sub>2g</sub> and A<sub>g</sub> modes in the Raman spectra of CaZrF<sub>6</sub>, from room temperature to about 200 °C. The same quantitative analysis cannot be done for the E<sub>g</sub> Raman band because of its broadening and weak intensity.

The top panel of Fig. 5 shows the Raman shift of the F<sub>2g</sub> and A<sub>g</sub> vibrational bands at three selected temperatures, while the bottom panel of the same figure shows the resulting temperature dependence of the vibrational frequencies. It is surprising to observe that the F<sub>2g</sub> mode, which is due to transverse vibrations and hence should be related to NTE, shows anomalous phonon hardening with increasing temperature. On the contrary the A<sub>g</sub> mode, assigned to stretching vibrations, undergoes a phonon softening with temperature. This is exactly the opposite of what one would expect, since the transverse vibrations (here the F<sub>2g</sub> mode), responsible for NTE, should display phonon softening with contracting crystal lattice, i.e., with temperature increasing.

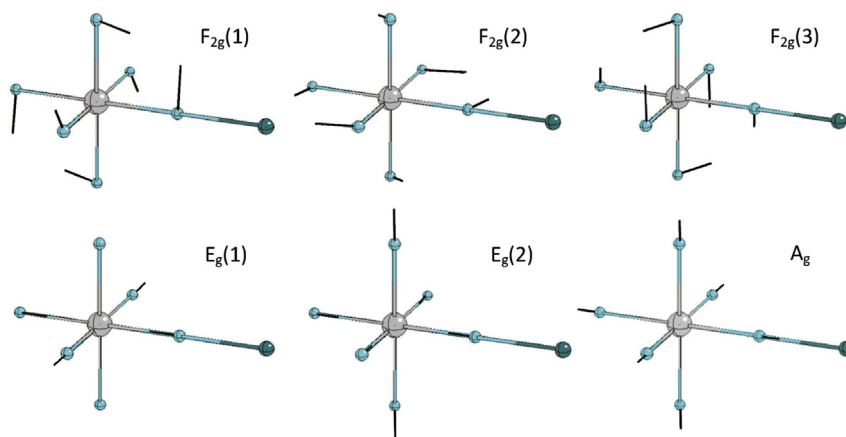
To explain this anomalous behavior, we need to consider that the temperature dependent shift  $\Delta\omega(T)$  of phonon frequency at constant pressure has two different contributions [29–31]: i) phonon frequency change  $\Delta\omega_{vol}(T)$  due to volume expansion, which is the “quasi-harmonic” contribution also called “implicit” anharmonicity, and ii) “explicit” or “true” anharmonic contribution  $\Delta\omega_{anh}(T)$ , which is the pure-temperature effect due to anharmonic phonon-phonon interactions, arising from cubic, quartic, and higher-order anharmonic terms in the interatomic potential. Therefore

$$\Delta\omega(T) = \Delta\omega_{vol}(T) + \Delta\omega_{anh}(T) \quad (2)$$

For an isotropic system, equation (2) in differential form becomes

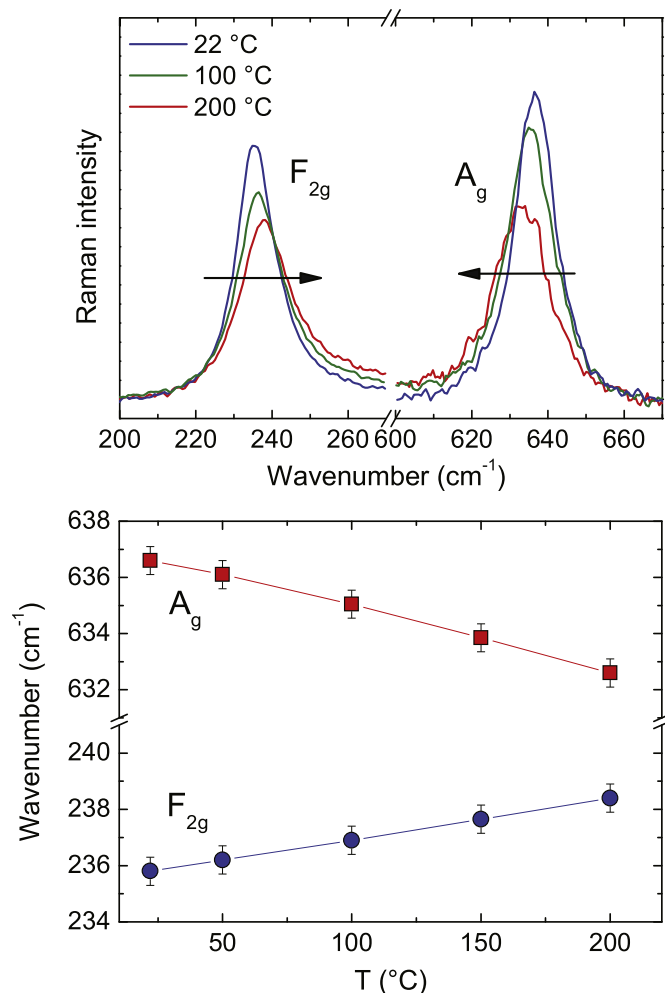
$$\left(\frac{\partial\omega_i}{\partial T}\right)_p = \left(\frac{\partial\omega_i}{\partial V}\right)_T \left(\frac{\partial V}{\partial T}\right)_p + \left(\frac{\partial\omega_i}{\partial T}\right)_V \quad (3)$$

which can also be written as



**Fig. 4.** Eigenvectors of the triple-degenerate F<sub>2g</sub> mode, of the double-degenerate E<sub>g</sub> mode, and of the A<sub>g</sub> mode (gray, green and cyan spheres represent Zr, Ca and F atoms, respectively). The F<sub>2g</sub> mode corresponds to bending vibrations of fluorine atoms perpendicular to the Zr-F-Ca bonds, leading to twisting of the ZrF<sub>6</sub> and CaF<sub>6</sub> octahedra. The E<sub>g</sub> and A<sub>g</sub> modes are, respectively, asymmetric and symmetric stretching of the Zr-F and Ca-F bonds in the ZrF<sub>6</sub> and CaF<sub>6</sub> octahedra. In particular, the A<sub>g</sub> mode corresponds to a breathing mode. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)





**Fig. 5.** Top panel: Raman shift of the  $F_{2g}$  and  $A_g$  vibrational bands at three selected temperatures. Bottom panel: temperature dependence of the vibrational frequency of the  $F_{2g}$  and  $A_g$  modes.

$$\frac{1}{\omega_i} \left( \frac{\partial \omega_i}{\partial T} \right)_p = -\alpha \gamma_i + \frac{1}{\omega_i} \left( \frac{\partial \omega_i}{\partial T} \right)_V \quad (4)$$

where  $\alpha$  is the coefficient of linear thermal expansion and  $\gamma_i$  is the isothermal Grüneisen parameter for the vibrational mode  $\omega_i$ . The term on the left-hand side of equation (4) gives the temperature-dependent frequency shift due to the total anharmonicity, as measured by temperature-dependent Raman spectra of Fig. 5. The first term on the right-hand side is the implicit anharmonicity due to volume contribution and can be determined from pressure-dependent frequency measurements and thermal expansion. The second term on the right-hand side is the explicit/true anharmonicity and cannot be determined experimentally. The temperature

dependence of the  $F_{2g}$  and  $A_g$  Raman modes shown in Fig. 5 is the result of the balance between implicit and explicit anharmonicity.

In order to separate implicit and explicit anharmonicity for the  $F_{2g}$  and  $A_g$  modes, we have calculated the total anharmonicity (Table 2) from the temperature-dependence of the Raman frequencies shown in the bottom panel of Fig. 5. The implicit anharmonicity has instead been estimated by thermal expansion ( $\alpha \approx -11.5 \times 10^{-6} \text{ K}^{-1}$  at 300 K [18]) and Grüneisen parameters  $\gamma_i$ , the latter obtained by pressure-dependent *ab initio* calculations performed in the pressure range from 0 to 300 MPa, corresponding to a reduction in lattice constant of about 0.15%. The resulting Grüneisen parameters for each vibrational modes at the Brillouin zone center are listed in the last column of Table 1, while the implicit anharmonic contribution of the  $F_{2g}$  and  $A_g$  Raman-active modes are reported in Table 2.

Firstly we find negative Grüneisen parameters for the lower-frequency modes (Table 1), in particular for the lowest-frequency  $F_{1g}$  symmetry mode corresponding to rigid rotation of  $\text{CaF}_6$  and  $\text{ZrF}_6$  octahedra around Ca and Zr, respectively, whose Grüneisen parameter is about  $-35$ . These findings are in agreement with the work of Hancock and co-workers [18] in which the most strongly negative Grüneisen parameters were associated with modes below  $\sim 150 \text{ cm}^{-1}$ ; furthermore, the large negative Grüneisen parameter of the  $F_{1g}$  mode indicates that rigid unit mode (RUM) mechanism [33,34] strongly contributes to the NTE in  $\text{CaZrF}_6$ . Note that typically the RUMs are optically active modes, like in  $\text{ZrW}_2\text{O}_8$  [31,32], whereas in  $\text{CaZrF}_6$  RUMs are silent due to the higher crystal symmetry. The linear thermal expansion coefficient  $\alpha$  of  $\text{CaZrF}_6$  at room temperature, calculated from the Grüneisen parameters listed in Table 1 (the reported value [18] of bulk modulus  $B=37 \text{ GPa}$  is used in the calculation), is  $\sim -2.3 \times 10^{-5} \text{ K}^{-1}$  versus an experimental value of  $\sim -11.5 \times 10^{-6} \text{ K}^{-1}$  [18]. Obviously there is an inherent limit in this thermal expansion calculation since the modes in Table 1 are only a fraction of the entire Brillouin zone.

Secondly, the explicit/true anharmonicity of the  $F_{2g}$  and  $A_g$  Raman-active modes of  $\text{CaZrF}_6$  has been determined by equation (4). As reported in Table 2, for both the  $F_{2g}$  and  $A_g$  modes the explicit anharmonic contribution is dominant over the implicit one. In addition, because the cubic anharmonic term leads to a decrease in frequency and, on the contrary, the quartic anharmonic term produces an increase in frequency [29,30], we can infer that cubic anharmonicity (i.e., three-phonon scattering processes) is dominant in the  $A_g$  phonon-mode, while the quartic anharmonicity (i.e., four-phonon scattering processes) is dominant in the  $F_{2g}$  phonon-mode. We should mention here that also in cubic  $\text{ScF}_3$ , which has the same  $\text{ReO}_3$ -type connectivity of  $\text{CaZrF}_6$ , there are a substantial number of modes with dominant quartic anharmonicity that accounts for phonon stiffening with temperature and for a significant part of the NTE [35,36]. As a result, the (negative) thermal expansion of  $\text{CaZrF}_6$  cannot be fully explained in the framework of quasi-harmonic approximation.

#### 4. Conclusions

The vibrational dynamics of NTE material  $\text{CaZrF}_6$  has been

**Table 2**

Anharmonic contributions (in  $10^{-5} \text{ K}^{-1}$ ) to the  $F_{2g}$  and  $A_g$  vibrational modes of  $\text{CaZrF}_6$ . The total anharmonicity has been determined from the temperature-dependence of the Raman frequencies shown in Fig. 5, the implicit anharmonicity has been estimated by pressure-dependent *ab initio* calculations and thermal expansion. The resulting explicit anharmonicity, separated using Eq. (4), indicates cubic anharmonicity (three-phonon processes) for the  $A_g$  mode and quartic anharmonicity (four-phonon processes) for the  $F_{2g}$  mode.

Modes	Total anharmonicity $1/\omega_i(\partial\omega_i/\partial T)_p$	Implicit anharmonicity $-\alpha\gamma_i$	Explicit anharmonicity $1/\omega_i(\partial\omega_i/\partial T)_V$
$F_{2g}$	6.18	-0.42	6.60
$A_g$	-3.81	1.45	-5.26

investigated by temperature-dependent Raman spectroscopy and *ab initio* calculations. The three expected Raman-active modes for CaZrF<sub>6</sub> have been reported for the first time. More interestingly, the vibrational frequency of the F<sub>2g</sub> and A<sub>g</sub> Raman modes, related to Ca-F-Zr bending and stretching vibrations, respectively, exhibit an anomalous dependence on temperature – exactly the opposite of what one would expect from the quasi-harmonic approximation. In order to shed light on this issue, we have separated implicit and explicit anharmonicity by means of pressure-dependent *ab initio* calculations. We have found out that explicit anharmonicity dominates over implicit anharmonicity, and the lower-frequency F<sub>2g</sub> mode has significant quartic anharmonicity, while the higher-frequency A<sub>g</sub> mode has significant cubic anharmonicity. The strong contribution of the explicit anharmonicity indicates that the large NTE of CaZrF<sub>6</sub> cannot be accurately predicted through the quasi-harmonic approximation.

## References

- [1] T.A. Mary, J.S.O. Evans, T. Vogt, A.W. Sleight, *Science* 272 (1996) 90.
- [2] P. Badrinarayanan, M.I. Ahmad, M. Akinc, M.R. Kessler, *Mater. Chem. Phys.* 131 (2011) 12.
- [3] A.L. Goodwin, C.J. Kepert, *Phys. Rev. B* 71 (2005) 140301 (R).
- [4] K.W. Chapman, P.J. Chupas, C.J. Kepert, *J. Am. Chem. Soc.* 127 (2005) 15630.
- [5] W. Zhou, H. Wu, T. Yildirim, J.R. Simpson, A.R.H. Walker, *Phys. Rev. B* 78 (2008) 054114.
- [6] B.K. Greve, K.L. Martin, P.L. Lee, P.J. Chupas, K.W. Chapman, A.P. Wilkinson, *J. Am. Chem. Soc.* 132 (2010) 15496.
- [7] J. Chen, K. Nittala, J.S. Forrester, J.L. Jones, J. Deng, R. Yu, X. Xing, *J. Am. Chem. Soc.* 133 (2011) 11114.
- [8] J. Chen, L. Fan, Y. Ren, Z. Pan, J. Deng, R. Yu, X. Xing, *Phys. Rev. Lett.* 110 (2013) 115901.
- [9] P. Mohn, *Nature* 400 (1999) 18.
- [10] J.R. Salvador, F. Gu, T. Hogan, M.G. Kanatzidis, *Nature* 425 (2003) 702.
- [11] J. Arvanitidis, K. Papagelis, S. Margadonna, K. Prassides, A.N. Fitch, *Nature* 425 (2003) 599.
- [12] G. Ernst, C. Broholm, G.R. Kowach, A.P. Ramirez, *Nature* 396 (1998) 147.
- [13] A. Sanson, F. Rocca, G. Dalba, P. Fornasini, R. Grisenti, M. Dapiaggi, G. Artioli, *Phys. Rev. B* 73 (2006) 214305.
- [14] A. Sanson, *Chem. Mater* 26 (2014) 3716.
- [15] J. Chen, L. Hu, J. Deng, X. Xing, *Chem. Soc. Rev.* 44 (2015) 3522.
- [16] L. Wang, P.F. Yuan, F. Wang, Q. Sun, Z.X. Guo, E.J. Liang, Y. Jia, *Mater. Chem. Phys.* 148 (2014) 214.
- [17] T.C. Ozawa, S.J. Kang, *J. Appl. Cryst.* 37 (2004) 679.
- [18] J.C. Hancock, K.W. Chapman, G.J. Halder, C.R. Morelock, B.S. Kaplan, L.C. Gallington, A. Bongiorno, C. Han, S. Zhou, A.P. Wilkinson, *Chem. Mater* 27 (2015) 3912.
- [19] R. Dovesi, R. Orlando, A. Erba, C.M. Zicovich-Wilson, B. Civalleri, S. Casassa, L. Maschio, M. Ferrabone, M. De La Pierre, P. Darco, Y. Noel, M. Causa, M. Rerat, B. Kirtman, *Int. J. Quantum Chem.* 114 (2014) 1287.
- [20] M. Catti, R. Dovesi, A. Pavese, V.R. Saunders, *J. Phys. Cond. Matter* 3 (1991) 4151.
- [21] R. Dovesi et al., <http://www.crystal.unito.it/Basis-Sets/zirconium.html> (unpublished).
- [22] R. Nada, C.R.A. Catlow, C. Pisani, R. Orlando, *Model. Simul. Mater. Sci. Eng.* 1 (1993) 165.
- [23] U. von Barth, L. Hedin, *J. Phys. C. Solid State Phys.* 5 (1972) 1629.
- [24] H.J. Monkhorst, J.D. Pack, *Phys. Rev. B* 13 (1976) 5188.
- [25] F. Pascale, C.M. Zicovich-Wilson, F.L. Gejo, B. Civalleri, R. Orlando, R. Dovesi, *J. Comput. Chem.* 25 (2004) 888.
- [26] C.M. Zicovich-Wilson, F. Pascale, C. Roetti, V.R. Saunders, R. Orlando, R. Dovesi, *J. Comput. Chem.* 25 (2004) 1873.
- [27] B.K. Kim, H. Hamaguchi, *Phys. Stat. Sol.(b)* 203 (1997) 557 (and references therein).
- [28] P.E. Quintard, P. Barb eris, A.P. Mirgorodsky, T. Merle-M ejean, *J. Am. Ceram. Soc.* 85 (2002) 1745.
- [29] G. Lucazeau, *J. Raman Spectr.* 34 (2003) 478.
- [30] N.P. Salke, M.K. Gupta, R. Rao, R. Mittal, J. Deng, X. Xing, *J. Appl. Phys.* 117 (2015) 235902.
- [31] T.R. Ravindran, A.K. Arora, T.A. Mary, *Phys. Rev. B* 67 (2003) 064301.
- [32] T.R. Ravindran, A.K. Arora, T.A. Mary, *J. Phys. Cond. Matter* 13 (2001) 11573.
- [33] J.Z. Tao, A.W. Sleight, *J. Sol. State Chem.* 173 (2003) 442.
- [34] M.T. Dove, K.O. Trachenko, M.G. Tucker, D.A. Keen, *Rev. Mineral. Geochem* 39 (2000) 1.
- [35] C.W. Li, X. Tang, J.A. Mun oz, J.B. Keith, S.J. Tracy, D.L. Abernathy, B. Fultz, *Phys. Rev. Lett.* 107 (2011) 195504.
- [36] P. Lazar, T. Bu cko, J. Hafner, *Phys. Rev. B* 92 (2015) 224302.